

(Photo: NH3 Tank—an icon of North American corn country)

Towards a more rational agriculture

The simplest and most powerful of all possible tests is the test of doing without.

Anhydrous ammonia (NH3), made up of one part nitrogen (N) and three parts hydrogen (H3), is one of the most widely used sources of nitrogen applied to keep crops green and prevent them from withering in the heat. Because NH3 contains no water—anhydrous = "without water"—and is therefore hydroscopic, it seeks moisture from the nearest source, including the human body. In its pressurized, liquid ammonia form, NH3 is injected about 15 cm under the soil's surface to prevent the liquid from vaporizing. The machinery used for this is complicated, hard to use and expensive. NH3 has a distinctly noxious ammonia smell which can easily be detected at distances of up to half a kilometer or more. (Humans can detect NH3 in concentrations as small as 5 parts per million (ppm); exposure to concentrations of 2,500 to 6,500 ppm can result in death.) Partly made from crude oil, NH3 has risen in price to about \$30 an acre, or about the profit that can be expected from the same acre of corn at current market prices. In addition to its agricultural applications, NH3 is widely stolen for the production of methamphetamine, known as "meth", a popular illegal street drug which affects the central nervous system and is strongly addictive.